Acta Crystallographica Section E Structure Reports Online

ISSN 1600-5368

N-(Anthracen-9-ylmethyl)adamantan-1amine

Wei-Qiang Fan^a* and Fu-Xiao Chen^b

^aSchool of Chemistry and Chemical Engineering, Jiangsu University, Zhenjiang 212023, People's Republic of China, and ^bSchool of the Environment, Jiangsu University, Zhenjiang 212023, People's Republic of China Correspondence e-mail: fwq4993329@ujs.edu.cn

Received 29 March 2012; accepted 16 May 2012

Key indicators: single-crystal X-ray study; T = 293 K; mean σ (C–C) = 0.002 Å; R factor = 0.039; wR factor = 0.104; data-to-parameter ratio = 15.1.

In the crystal stucture of the of the title compound, $C_{25}H_{27}N$, stong π - π interactions are found between adjacent anthracene fragments, with a shortest centroid–centroid distance of 3.5750 (9) Å.

Related literature

Anthracene derivatives have been widely used in the field of anion recognition, metal ion fluorescent sensors, as well as pH sensors, see: Bernhardt *et al.* (2001), Chen & Chen (2004); Gunnlaugsson *et al.* (2003); Kim & Yoon (2002)



Experimental

Crystal data

 $C_{25}H_{27}N$ $M_r = 341.26$ Orthorhombic, *Pccn a* = 9.9546 (4) Å *b* = 42.1921 (19) Å *c* = 8.6133 (4) Å

Data collection

Bruker APEX CCD area-detector diffractometer Absorption correction: multi-scan (*SADABS*; Bruker, 1999) $T_{\rm min} = 0.975, T_{\rm max} = 0.998$

Refinement

 $R[F^2 > 2\sigma(F^2)] = 0.039$ $wR(F^2) = 0.104$ S = 1.03 3594 reflections 238 parameters 1 restraint $V = 3617.6 \text{ (3) } \text{\AA}^{3}$ Z = 8 Mo K\alpha radiation $\mu = 0.07 \text{ mm}^{-1}$ T = 293 K 0.35 \times 0.24 \times 0.20 mm

18689 measured reflections 3594 independent reflections 3089 reflections with $I > 2\sigma(I)$ $R_{\text{int}} = 0.037$

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL-Plus* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97*.

We thank the Start-up Foundation for Advanced Professionals of Jiangsu University (11JDG105) for support.

Supplementary data and figures for this paper are available from the IUCr electronic archives (Reference: AA2056).

References

Bernhardt, P. V., Moore, E. G. & Riley, M. J. (2001). *Inorg. Chem.* 40, 5799–5805.

Bruker (1999). SADABS, SMART and SAINT. Bruker AXS Inc., Madison, Wisconsin, USA.

Chen, Q. Y. & Chen, C. F. (2004). Tetrahedron Lett. 45, 6493-6496.

Gunnlaugsson, T., Lee, T. C. & Parkesh, R. (2003). Org. Lett. 5, 4065–4068. Kim, S. K. & Yoon, J. (2002). Chem. Commun. pp. 770–771.

Sheldrick, G. M. (2008). Acta Cryst. A64, 112-122.

supplementary materials

Acta Cryst. (2012). E68, o1831 [doi:10.1107/S1600536812022106]

N-(Anthracen-9-ylmethyl)adamantan-1-amine

Wei-Qiang Fan and Fu-Xiao Chen

Comment

Anthracene derivatives have been widely used in the field of anion recognition, metal ionfluorescent sensors, as well as pH sensors (Gunnlaugsson *et al.*, 2003; Chen & Chen, 2004; Kim & Yoon, 2002; Bernhardt *et al.* 2001) because of their excellent photophysical properties and high fluorescence.

In the crystals of the title compound (Fig. 1), there are two π - π interactions between benzene rings of the ajacent anthracene fragments with the distances $Cg_1 \cdots Cg_2^i = 3.5750$ (9) Å and $Cg_1 \cdots Cg_3^i = 4.0043$ (10) Å. (Cg_1 , Cg_2 and Cg_3 are the centroids of the rings [C1/C5 and C14], [C5/C7 and C12/C14] and [C7/C12], respectively; symmetry code: (i) 1/2 - x, y, z + 1/2) forming one-dimensional supramolecular chains along c axis direction (Fig. 2).

Experimental

9-Anthracenecarboxaldehyde (2.06 g, 10 mmol) was added into a solution of amantadine (1.51 g, 10 mmol) in ethanol.
Yellow precipitate was formed atfer string for 1 h. The yellow Schiff base was filtrated and dryed. NaBH₄ (7.56 g, 20 mmol) was added into a solution of the Schiff base in anhydrous methanol (120 ml). After 3 h, the white solid,
9-[(adamantan-1-ylamino)methyl]anthracene, was obtained by reduced pressure distillation, extraction and drying. The colourless block-shaped crystals of the title compound suitable for X-ray analysis were obtained by recrystallization from ethanol.

Refinement

H atom bonded to N was located in a difference Fourier map and refined isotropically with a bond restraint of N—H= 0.85 Å and $U_{iso}(H) = 1.5 U_{eq}(N)$. Other H atoms were placed in calculated positions with C—H distances 0.93 (aromatic), 0.97 Å (methylene) and 0.97 Å (methine) and refined as riding with $U_{iso}(H) = 1.2U_{eq}(C)$.

Computing details

Data collection: *SMART* (Bruker, 1999); cell refinement: *SAINT* (Bruker, 1999); data reduction: *SAINT* (Bruker, 1999); program(s) used to solve structure: *SHELXS97* (Sheldrick, 2008); program(s) used to refine structure: *SHELXL97* (Sheldrick, 2008); molecular graphics: *SHELXTL-Plus* (Sheldrick, 2008); software used to prepare material for publication: *SHELXL97* (Sheldrick, 2008).



Figure 1

The molecular structure of the title compound. Thermal displacement ellipsoids are drawn at the 30% probability level.



Figure 2

The one-dimensional supramolecular chains linked by π - π interactions. H atoms are omitted for clarity.

N-(Anthracen-9-ylmethyl)adamantan-1-amine

Crystal data	
$C_{25}H_{27}N$	F(000) = 1472
$M_r = 341.26$	$D_{\rm x} = 1.253 {\rm ~Mg} {\rm ~m}^{-3}$
Orthorhombic, Pccn	Mo <i>K</i> α radiation, $\lambda = 0.71073$ Å
Hall symbol: -P 2ab 2ac	Cell parameters from 3594 reflections
a = 9.9546 (4) Å	$\theta = 1.0-26.0^{\circ}$
b = 42.1921 (19) Å	$\mu=0.07~\mathrm{mm}^{-1}$
c = 8.6133 (4) Å	T = 293 K
V = 3617.6 (3) Å ³	Block, colorless
Z = 8	$0.35 \times 0.24 \times 0.20 \text{ mm}$
Data collection	
Bruker APEX CCD area-detector	Absorption correction: multi-scan

diffractometer Radiation source: fine-focus sealed tube Graphite monochromator ω scans Absorption correction: multi-scan (*SADABS*; Bruker, 1999) $T_{min} = 0.975$, $T_{max} = 0.998$ 18689 measured reflections 3594 independent reflections 3089 reflections with $I > 2\sigma(I)$

$R_{\rm int} = 0.037$	$k = -52 \rightarrow 50$
$\theta_{\rm max} = 26.0^{\circ}, \theta_{\rm min} = 1.0^{\circ}$	$l = -10 \rightarrow 10$
$h = -8 \rightarrow 12$	

Refinement	
Refinement on F^2	Secondary atom site location: difference Fourier
Least-squares matrix: full	map
$R[F^2 > 2\sigma(F^2)] = 0.039$	Hydrogen site location: inferred from
$wR(F^2) = 0.104$	neighbouring sites
<i>S</i> = 1.03	H atoms treated by a mixture of independent
3594 reflections	and constrained refinement
238 parameters	$w = 1/[\sigma^2(F_o^2) + (0.0431P)^2 + 1.8749P]$
1 restraint	where $P = (F_o^2 + 2F_c^2)/3$
Primary atom site location: structure-invariant	$(\Delta/\sigma)_{\rm max} < 0.001$
direct methods	$\Delta \rho_{\rm max} = 0.30 \text{ e } \text{\AA}^{-3}$
	$\Delta \rho_{\rm min} = -0.20 \text{ e } \text{\AA}^{-3}$

Special details

Geometry. All s.u.'s (except the s.u. in the dihedral angle between two l.s. planes) are estimated using the full covariance matrix. The cell s.u.'s are taken into account individually in the estimation of s.u.'s in distances, angles and torsion angles; correlations between s.u.'s in cell parameters are only used when they are defined by crystal symmetry. An approximate (isotropic) treatment of cell s.u.'s is used for estimating s.u.'s involving l.s. planes.

Refinement. Refinement of F^2 against ALL reflections. The weighted *R*-factor *wR* and goodness of fit *S* are based on F^2 , conventional *R*-factors *R* are based on *F*, with *F* set to zero for negative F^2 . The threshold expression of $F^2 > \sigma(F^2)$ is used only for calculating *R*-factors(gt) *etc.* and is not relevant to the choice of reflections for refinement. *R*-factors based on F^2 are statistically about twice as large as those based on *F*, and *R*- factors based on ALL data will be even larger.

	x	У	Ζ	$U_{ m iso}$ */ $U_{ m eq}$	
C1	0.07687 (14)	0.06244 (3)	0.02106 (16)	0.0213 (3)	
H1	0.0190	0.0763	0.0720	0.026*	
C2	0.02463 (14)	0.03905 (3)	-0.06924 (16)	0.0251 (3)	
H2	-0.0681	0.0372	-0.0795	0.030*	
C3	0.10948 (15)	0.01740 (3)	-0.14794 (16)	0.0263 (3)	
H3	0.0724	0.0015	-0.2089	0.032*	
C4	0.24486 (15)	0.01993 (3)	-0.13433 (15)	0.0239 (3)	
H4	0.2998	0.0057	-0.1866	0.029*	
C5	0.30441 (14)	0.04417 (3)	-0.04106 (15)	0.0189 (3)	
C6	0.44299 (14)	0.04664 (3)	-0.02481 (15)	0.0206 (3)	
H6	0.4980	0.0321	-0.0753	0.025*	
C7	0.50153 (13)	0.07034 (3)	0.06532 (15)	0.0192 (3)	
C8	0.64357 (14)	0.07199 (3)	0.08598 (18)	0.0275 (3)	
H8	0.6984	0.0572	0.0370	0.033*	
C9	0.70002 (15)	0.09479 (3)	0.1760 (2)	0.0320 (4)	
H9	0.7925	0.0951	0.1911	0.038*	
C10	0.61811 (14)	0.11826 (3)	0.24708 (17)	0.0266 (3)	
H10	0.6577	0.1342	0.3061	0.032*	
C11	0.48252 (13)	0.11763 (3)	0.22964 (15)	0.0201 (3)	
H11	0.4309	0.1334	0.2755	0.024*	
C12	0.41699 (13)	0.09318 (3)	0.14216 (14)	0.0165 (3)	
C13	0.27606 (13)	0.09040 (3)	0.13158 (14)	0.0164 (3)	

Fractional atomic coordinates and isotropic or equivalent isotropic displacement parameters $(Å^2)$

C14	0.21886 (13)	0.06624 (3)	0.03960 (14)	0.0177 (3)
C15	0.18568 (13)	0.11250 (3)	0.22289 (14)	0.0173 (3)
H15A	0.1067	0.1010	0.2570	0.021*
H15B	0.2330	0.1199	0.3145	0.021*
C16	0.07339 (12)	0.16531 (3)	0.21643 (14)	0.0148 (3)
C17	-0.03739 (12)	0.15278 (3)	0.32484 (14)	0.0161 (3)
H17A	-0.1027	0.1409	0.2647	0.019*
H17B	0.0018	0.1385	0.4007	0.019*
C18	-0.10802 (12)	0.18021 (3)	0.40888 (15)	0.0178 (3)
H18	-0.1777	0.1717	0.4777	0.021*
C19	-0.17216 (13)	0.20239 (3)	0.28916 (16)	0.0211 (3)
H19A	-0.2167	0.2198	0.3417	0.025*
H19B	-0.2389	0.1909	0.2293	0.025*
C20	-0.06276 (13)	0.21533 (3)	0.18062 (16)	0.0204 (3)
H20	-0.1037	0.2294	0.1034	0.024*
C21	0.04073 (13)	0.23400 (3)	0.27656 (16)	0.0214 (3)
H21A	0.1096	0.2425	0.2087	0.026*
H21B	-0.0030	0.2516	0.3287	0.026*
C22	0.10463 (13)	0.21183 (3)	0.39676 (15)	0.0193 (3)
H22	0.1706	0.2237	0.4581	0.023*
C23	0.17531 (12)	0.18428 (3)	0.31292 (15)	0.0172 (3)
H23A	0.2450	0.1926	0.2454	0.021*
H23B	0.2174	0.1705	0.3887	0.021*
C24	0.00819 (13)	0.18773 (3)	0.09765 (15)	0.0185 (3)
H24A	-0.0565	0.1761	0.0357	0.022*
H24B	0.0766	0.1960	0.0285	0.022*
C25	-0.00415 (13)	0.19856 (3)	0.50496 (15)	0.0191 (3)
H25A	0.0364	0.1846	0.5811	0.023*
H25B	-0.0479	0.2158	0.5598	0.023*
N1	0.14369 (11)	0.14005 (2)	0.12914 (12)	0.0168 (2)
H1N	0.0920 (15)	0.1338 (3)	0.0533 (16)	0.025*

Atomic displacement parameters $(Å^2)$

	U^{11}	U^{22}	U^{33}	U^{12}	U^{13}	U^{23}
C1	0.0250 (7)	0.0178 (6)	0.0210 (7)	0.0004 (5)	0.0014 (5)	0.0045 (5)
C2	0.0253 (7)	0.0248 (7)	0.0253 (7)	-0.0064 (6)	-0.0025 (6)	0.0061 (6)
C3	0.0386 (8)	0.0200 (7)	0.0204 (7)	-0.0101 (6)	-0.0025 (6)	0.0001 (5)
C4	0.0371 (8)	0.0158 (6)	0.0189 (7)	-0.0018 (6)	0.0031 (6)	-0.0011 (5)
C5	0.0282 (7)	0.0130 (6)	0.0155 (6)	-0.0009 (5)	0.0019 (5)	0.0020 (5)
C6	0.0272 (7)	0.0143 (6)	0.0204 (7)	0.0037 (5)	0.0065 (5)	-0.0012 (5)
C7	0.0231 (7)	0.0146 (6)	0.0199 (7)	0.0013 (5)	0.0042 (5)	0.0016 (5)
C8	0.0223 (7)	0.0218 (7)	0.0384 (9)	0.0026 (5)	0.0077 (6)	-0.0037 (6)
C9	0.0195 (7)	0.0279 (7)	0.0488 (10)	-0.0032 (6)	0.0031 (7)	-0.0058 (7)
C10	0.0268 (7)	0.0208 (7)	0.0323 (8)	-0.0067 (5)	0.0014 (6)	-0.0042 (6)
C11	0.0257 (7)	0.0146 (6)	0.0201 (7)	0.0000 (5)	0.0038 (5)	0.0003 (5)
C12	0.0229 (6)	0.0119 (6)	0.0147 (6)	0.0009 (5)	0.0024 (5)	0.0022 (5)
C13	0.0231 (7)	0.0120 (5)	0.0139 (6)	0.0021 (5)	0.0023 (5)	0.0041 (5)
C14	0.0240 (7)	0.0138 (6)	0.0152 (6)	0.0004 (5)	0.0017 (5)	0.0044 (5)
C15	0.0203 (6)	0.0151 (6)	0.0165 (6)	0.0013 (5)	0.0021 (5)	0.0015 (5)

C16	0.0168 (6)	0.0124 (6)	0.0153 (6)	0.0010 (5)	0.0002 (5)	-0.0002 (5)	
C17	0.0173 (6)	0.0141 (6)	0.0170 (6)	-0.0018 (5)	0.0009 (5)	-0.0005 (5)	
C18	0.0176 (6)	0.0165 (6)	0.0192 (6)	-0.0003 (5)	0.0038 (5)	-0.0013 (5)	
C19	0.0179 (6)	0.0209 (6)	0.0246 (7)	0.0044 (5)	-0.0008 (5)	-0.0036 (5)	
C20	0.0249 (7)	0.0156 (6)	0.0207 (7)	0.0052 (5)	-0.0020 (5)	0.0027 (5)	
C21	0.0259 (7)	0.0129 (6)	0.0254 (7)	0.0006 (5)	0.0025 (6)	0.0008 (5)	
C22	0.0206 (6)	0.0146 (6)	0.0226 (7)	-0.0029 (5)	-0.0024 (5)	-0.0028 (5)	
C23	0.0158 (6)	0.0153 (6)	0.0206 (7)	-0.0008(5)	-0.0005 (5)	0.0016 (5)	
C24	0.0215 (6)	0.0180 (6)	0.0160 (6)	0.0020 (5)	-0.0006 (5)	0.0012 (5)	
C25	0.0256 (7)	0.0145 (6)	0.0171 (6)	0.0019 (5)	-0.0006 (5)	-0.0022 (5)	
N1	0.0212 (6)	0.0145 (5)	0.0147 (5)	0.0037 (4)	-0.0003 (4)	0.0002 (4)	
C22 C23 C24 C25 N1	0.0206 (6) 0.0158 (6) 0.0215 (6) 0.0256 (7) 0.0212 (6)	0.0146 (6) 0.0153 (6) 0.0180 (6) 0.0145 (6) 0.0145 (5)	0.0226 (7) 0.0206 (7) 0.0160 (6) 0.0171 (6) 0.0147 (5)	-0.0029 (5) -0.0008 (5) 0.0020 (5) 0.0019 (5) 0.0037 (4)	-0.0024 (5) -0.0005 (5) -0.0006 (5) -0.0006 (5) -0.0003 (4)	-0.0028 (5) 0.0016 (5) 0.0012 (5) -0.0022 (5) 0.0002 (4)	

Geometric parameters (Å, °)

C1—C2	1.3598 (19)	C16—N1	1.4800 (15)
C1—C14	1.4314 (19)	C16—C23	1.5364 (16)
C1—H1	0.9300	C16—C24	1.5373 (16)
C2—C3	1.417 (2)	C16—C17	1.5387 (16)
С2—Н2	0.9300	C17—C18	1.5355 (16)
C3—C4	1.357 (2)	C17—H17A	0.9700
С3—Н3	0.9300	C17—H17B	0.9700
C4—C5	1.4291 (18)	C18—C19	1.5318 (18)
C4—H4	0.9300	C18—C25	1.5339 (17)
C5—C6	1.3905 (19)	C18—H18	0.9800
C5—C14	1.4407 (18)	C19—C20	1.5357 (18)
С6—С7	1.3937 (18)	C19—H19A	0.9700
С6—Н6	0.9300	C19—H19B	0.9700
C7—C8	1.4268 (19)	C20—C24	1.5378 (17)
C7—C12	1.4403 (17)	C20—C21	1.5379 (18)
C8—C9	1.357 (2)	C20—H20	0.9800
С8—Н8	0.9300	C21—C22	1.5335 (18)
C9—C10	1.422 (2)	C21—H21A	0.9700
С9—Н9	0.9300	C21—H21B	0.9700
C10—C11	1.3584 (19)	C22—C25	1.5346 (18)
С10—Н10	0.9300	C22—C23	1.5389 (17)
C11—C12	1.4346 (17)	C22—H22	0.9800
C11—H11	0.9300	C23—H23A	0.9700
C12—C13	1.4107 (18)	C23—H23B	0.9700
C13—C14	1.4110 (17)	C24—H24A	0.9700
C13—C15	1.5155 (16)	C24—H24B	0.9700
C15—N1	1.4760 (15)	C25—H25A	0.9700
C15—H15A	0.9700	C25—H25B	0.9700
C15—H15B	0.9700	N1—H1N	0.873 (13)
C2—C1—C14	121.52 (13)	C16—C17—H17A	109.5
C2-C1-H1	119.2	C18—C17—H17B	109.5
C14—C1—H1	119.2	C16—C17—H17B	109.5
C1—C2—C3	120.90 (13)	H17A—C17—H17B	108.1
С1—С2—Н2	119.6	C19—C18—C25	109.63 (10)
С3—С2—Н2	119.5	C19—C18—C17	109.51 (10)

C_{4} C_{3} C_{2}	120.00(12)	C25-C18-C17	109.03(10)
C4-C3-H3	120.00 (12)	C19 - C18 - H18	109.05 (10)
С2—С3—Н3	120.0	C_{25} C_{18} H_{18}	109.6
$C_2 C_3 C_4 C_5$	120.0	C_{12}^{12} C_{18}^{10} H_{18}^{10}	109.6
$C_3 = C_4 = C_3$	121.12 (13)	$C_{18}^{18} C_{19}^{19} C_{20}^{20}$	109.0
$C_5 = C_4 = H_4$	119.4	$C_{18} = C_{19} = C_{20}$	109.55 (10)
C_{5} C_{4} C_{4}	119.4 121 45 (12)	$C_{10} = C_{10} = H_{10A}$	109.8
C6 - C5 - C4	121.43(12) 110.20(12)	C_{20} C_{19} H_{10} H_{10}	109.8
$C_{0} - C_{3} - C_{14}$	119.30(12)	С18—С19—Н19В	109.8
C4 - C3 - C14	119.24 (12)	C20—C19—H19B	109.8
C_{2}	121.65 (12)	HI9A—CI9—HI9B	108.3
С5—С6—Н6	119.2	C19 - C20 - C24	109.84 (10)
С/—С6—Н6	119.2	C19—C20—C21	109.27 (11)
C6-C/-C8	121.25 (12)	C24—C20—C21	109.26 (10)
C6—C7—C12	119.45 (12)	С19—С20—Н20	109.5
C8—C7—C12	119.29 (12)	C24—C20—H20	109.5
C9—C8—C7	121.07 (13)	C21—C20—H20	109.5
С9—С8—Н8	119.5	C22—C21—C20	109.15 (10)
С7—С8—Н8	119.5	C22—C21—H21A	109.8
C8—C9—C10	120.16 (13)	C20—C21—H21A	109.8
С8—С9—Н9	119.9	C22—C21—H21B	109.8
С10—С9—Н9	119.9	C20—C21—H21B	109.8
C11—C10—C9	120.58 (13)	H21A—C21—H21B	108.3
C11—C10—H10	119.7	C21—C22—C25	109.87 (11)
С9—С10—Н10	119.7	C21—C22—C23	109.49 (10)
C10-C11-C12	121.59 (12)	C25—C22—C23	109.38 (10)
C10-C11-H11	119.2	C21—C22—H22	109.4
C12—C11—H11	119.2	C25—C22—H22	109.4
C13—C12—C11	123.08 (11)	C23—C22—H22	109.4
C13—C12—C7	119.73 (11)	C16—C23—C22	110.21 (10)
C11—C12—C7	117.18 (12)	C16—C23—H23A	109.6
C12—C13—C14	119.83 (11)	С22—С23—Н23А	109.6
C12—C13—C15	120.39 (11)	C16—C23—H23B	109.6
C14—C13—C15	119.75 (11)	С22—С23—Н23В	109.6
C13—C14—C1	122.83 (12)	H23A—C23—H23B	108.1
C13—C14—C5	119.96 (12)	C16—C24—C20	110.53 (10)
C1—C14—C5	117.21 (12)	C16—C24—H24A	109.5
N1—C15—C13	111.64 (10)	C20—C24—H24A	109.5
N1—C15—H15A	109.3	C16—C24—H24B	109.5
C13—C15—H15A	109.3	C20—C24—H24B	109.5
N1-C15-H15B	109.3	H24A—C24—H24B	108.1
C13—C15—H15B	109.3	C18 - C25 - C22	109 40 (10)
H15A - C15 - H15B	108.0	C18 - C25 - H25A	109.10 (10)
N1-C16-C23	109.73 (10)	C^{22} C^{25} H^{25A}	109.8
N1-C16-C24	107.74(10)	C18 - C25 - H25R	109.8
C_{23} C_{16} C_{24}	108 55 (10)	C_{22} C_{25} H_{25B}	109.8
N1-C16-C17	113 57 (9)	$H_{25} = C_{25} = H_{25} = H_{25}$	102.0
C_{23} C_{16} C_{17}	108.90 (10)	C15 - N1 - C16	115 03 (0)
C_{24} C_{16} C_{17}	108.22 (10)	C15 $N1$ $H1N$	109.8 (10)
$C_{24} = C_{10} = C_{17}$	100.22(10) 110.81(10)	C_{10} M_{10} M	109.0(10) 108.6(10)
-10 - 01/ - 010	110.01 (10)	U10-111-1111N	100.0(10)

C18—C17—H17A	109.5		
C14—C1—C2—C3	0.3 (2)	C12—C13—C15—N1	94.66 (13)
C1—C2—C3—C4	-0.3 (2)	C14—C13—C15—N1	-87.49 (13)
C2—C3—C4—C5	0.2 (2)	N1-C16-C17-C18	-178.78 (10)
C3—C4—C5—C6	179.09 (13)	C23—C16—C17—C18	58.63 (13)
C3—C4—C5—C14	-0.04 (19)	C24—C16—C17—C18	-59.20 (13)
C4—C5—C6—C7	179.63 (12)	C16—C17—C18—C19	60.18 (13)
C14—C5—C6—C7	-1.24 (19)	C16—C17—C18—C25	-59.77 (13)
C5—C6—C7—C8	177.65 (13)	C25—C18—C19—C20	60.30 (13)
C5—C6—C7—C12	-0.85 (19)	C17—C18—C19—C20	-59.28 (13)
C6—C7—C8—C9	-178.87 (14)	C18—C19—C20—C24	59.26 (13)
С12—С7—С8—С9	-0.4 (2)	C18—C19—C20—C21	-60.59 (13)
C7—C8—C9—C10	-2.3 (2)	C19—C20—C21—C22	60.28 (13)
C8—C9—C10—C11	1.9 (2)	C24—C20—C21—C22	-59.93 (13)
C9—C10—C11—C12	1.2 (2)	C20—C21—C22—C25	-59.94 (13)
C10-C11-C12-C13	174.98 (13)	C20—C21—C22—C23	60.21 (13)
C10-C11-C12-C7	-3.74 (19)	N1-C16-C23-C22	176.69 (10)
C6—C7—C12—C13	3.06 (18)	C24—C16—C23—C22	59.19 (13)
C8—C7—C12—C13	-175.46 (12)	C17—C16—C23—C22	-58.43 (12)
C6—C7—C12—C11	-178.18 (11)	C21—C22—C23—C16	-60.42 (13)
C8—C7—C12—C11	3.30 (18)	C25—C22—C23—C16	60.03 (13)
C11—C12—C13—C14	178.18 (11)	N1-C16-C24-C20	-178.02 (10)
C7—C12—C13—C14	-3.14 (17)	C23—C16—C24—C20	-59.25 (13)
C11—C12—C13—C15	-3.97 (18)	C17—C16—C24—C20	58.80 (13)
C7—C12—C13—C15	174.71 (11)	C19—C20—C24—C16	-59.76 (13)
C12—C13—C14—C1	-178.69 (11)	C21—C20—C24—C16	60.10 (13)
C15—C13—C14—C1	3.44 (17)	C19—C18—C25—C22	-59.63 (13)
C12—C13—C14—C5	1.06 (17)	C17—C18—C25—C22	60.24 (13)
C15—C13—C14—C5	-176.81 (11)	C21—C22—C25—C18	59.59 (13)
C2-C1-C14-C13	179.54 (12)	C23—C22—C25—C18	-60.63 (13)
C2-C1-C14-C5	-0.22 (18)	C13—C15—N1—C16	-170.68 (10)
C6—C5—C14—C13	1.15 (18)	C23—C16—N1—C15	75.36 (12)
C4—C5—C14—C13	-179.70 (11)	C24—C16—N1—C15	-166.62 (10)
C6-C5-C14-C1	-179.09 (11)	C17—C16—N1—C15	-46.77 (14)
C4—C5—C14—C1	0.06 (17)		